

tioned in this communication. A poor yield of the base (VII) is obtained by condensation of the ketone with 6-aminopiperonal.

EXPERIMENTAL.

7-Methoxychromanone (compare Perkin, Rây, and Robinson, J., 1926, 945; 1927, 2097).—The ring-closure of β -*m*-methoxyphenoxypropionic acid was studied under a graded series of conditions and the best yield (62%) was obtained by the use of 75% sulphuric acid.

Methoxyphenoxypropionic acid (3.5 g.) was dissolved in 75% sulphuric acid (25 c.c., *d* 1.677/19°) and the mixture was heated during 2 hours to 60—70° and then during 4 hours to 90°. The product was isolated in the known manner. When 78% sulphuric acid was used, the yield dropped to 53% (80% acid, 48%; 82%, 45%; 85%, 42%). We think that it will be worth while to try sulphuric acid of even a lower concentration than 75%.

7-Hydroxy-2 : 3-[7'-methoxychromeno(4' : 3')]benzopyrylium Chloride.—When hydrogen chloride was passed through a saturated solution of 7-methoxychromanone and β -resorcylaldehyde in acetic acid, the resulting yield of *pyrylium* salt was 15%. Under the following conditions the yield was almost theoretical. A solution of 7-methoxychromanone (3.6 g.) and β -resorcylaldehyde (2.8 g.) in dry ethyl acetate (30 c.c.) was saturated with hydrogen chloride during 4 hours and the mixture was later maintained at 25—30° for 2 hours and kept at room temperature for 24 hours. The separated crystals (yield, 95%), recrystallised from 5% hydrochloric acid, formed slender red needles (Found in material dried at 50° in a vacuum : C, 61.3; H, 4.5. $C_{17}H_{13}O_4Cl \cdot H_2O$ requires C, 61.0; H, 4.5. Found in material dried at 110—120° : C, 64.3; H, 4.2. $C_{17}H_{13}O_4Cl$ requires C, 64.4; H, 4.1%). The salt is sparingly soluble in acetic acid, readily in alcohol; all its reddish-orange acid solutions exhibit a strong, greenish-yellow fluorescence. The quinone base obtained by the action of sodium acetate has the typical blood-red colour of its class; it is sparingly soluble in benzene to a deep red solution and is reconverted by 5% hydrochloric acid into the orange-red *pyrylium* salt.

The catalytic reduction of this salt presented unexpected difficulty. When an active platinum catalyst was used in acetic acid solution, only one molecule of hydrogen was absorbed and aerial re-oxidation to the *pyrylium* salt occurred readily. By using zinc dust and acetic acid, reduction was brought about, but the product, possibly a dimerised compound, could not be crystallised.

8-Methoxy-2 : 3-[7'-methoxychromeno(4' : 3')]benzopyrylium Ferrichloride (V).—When *o*-vanillin is substituted for β -resorcylaldehyde, an equally facile condensation occurs; the resulting chloride is soluble in dilute aqueous acids to orange-red solutions which do not exhibit fluorescence. The *ferrichloride*, obtained in the usual way, crystallised from acetic acid in well-formed orange needles, m. p. 202° (unsharp) (Found : C, 44.0; H, 3.0. $C_{18}H_{15}O_4Cl_4Fe$ requires C, 43.9; H, 3.1%).

7-Hydroxy-5-benzoyloxy-2 : 3-[7'-methoxychromeno(4' : 3')]benzopyrylium Chloride.—This salt was obtained, like the derivative from β -resorcylaldehyde, by the use of 2-*O*-benzoylphloroglucinaldehyde (yield, 72%). It is very sparingly soluble in hot dilute hydrochloric acid to a red solution exhibiting a yellow fluorescence and it is readily soluble in alcohol to an intensely red solution. It crystallised from a mixture of equal volumes of alcohol and 10% aqueous hydrochloric acid in very slender, red, microscopic needles (Found in material dried at 80° in a vacuum : C, 66.2; H, 4.3. $C_{24}H_{17}O_6Cl$ requires C, 66.0; H, 3.9%).

6 : 7-Dimethoxy-2 : 3-[7'-methoxychromeno(4' : 3')]benzopyrylium Ferrichloride (II).—The condensation of 7-methoxychromanone and 2-hydroxy-4 : 5-dimethoxybenzaldehyde (for a specimen of which we are greatly indebted to Professor A. Robertson) was carried out in ethyl acetate saturated with hydrogen chloride at 0°, the mixture being kept for 6 days. An almost theoretical yield was obtained of a bright red chloride, which was converted into the *ferrichloride* in the usual way. This characteristic substance crystallised from acetic acid, in which it was sparingly soluble, in crimson elongated prisms with a blue reflex (Found : C, 43.5; H, 3.3. Calc. for $C_{18}H_{15}O_5Cl_4Fe$: C, 43.6; H, 3.3%). The substance had m. p. 213° (decomp.), alone or mixed with a specimen made by the method of Perkin, Rây, and Robinson (*loc. cit.*), and direct comparison confirmed the identity.

6 : 7-Methylenedioxy-2 : 3-[7'-methoxychromeno(4' : 3')]quinoline (VII) (*Picrate*).—An alcoholic solution of 7-methoxychromanone and 6-aminopiperonal, together with a small relative volume of 20% aqueous potassium hydroxide, was refluxed for 8 hours; acetic acid in excess was then added, and the boiling continued for 3 hours. The base was isolated as the *picrate*, which crystallised from acetone in slender needles, m. p. ca. 244° (Found in material dried at 80° in

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a vacuum: C, 54.0; H, 3.4; N, 10.5. $C_{18}H_{13}O_4N, C_6H_3O_7N_3$ requires C, 53.8; H, 3.0; N, 10.5%). The hydrochloride was prepared for physiological investigation by decomposition of the picrate in methyl-alcoholic hydrogen chloride and precipitation with ether; its solutions exhibit a bright blue fluorescence.

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